

CHEMISTRY

TARGET: JEE Advanced - 2023

CAPS - 9

Hydrocarbons Part-I

1.
$$\xrightarrow{\text{Acid}}$$
 Major product is -

2. Ethyl benzene
$$Cl_2$$
 Fe P NBS Q $leq. Li/THF$ $leq. Li/THF$ $Oxirane$ R

[S] is -

(D) None of these

3.

$$CH_3$$
 CH_3
 CH_3CI
 $AICI_3$
 CH_3CI

Most probable structure of P is:

(A)
$$_{\mathrm{H_3C}}$$
 (B) $_{\mathrm{CN}}$ (C) $_{\mathrm{CN}}$ (C) $_{\mathrm{CH_3}}$

(D) Reaction will not take place due to strong deactivating nature of -CN

4. The major product in the reaction

S is -

5. Which of the following aromatic ring undergoes fastest D exchange with DF / BF₃.

(A)
$$CH_2$$
 Br Br $COOH$ NH_2 Br Br

C)
$$CH_2$$
 Br
 $CD) H_2N$
 $COOH NH_2$

7. Which steps is used to produce 1–Chloro–3–ethylbenzene

$$(A) \bigcirc \xrightarrow{CH_3-C-O-C-CH_3} \xrightarrow{Cl_2/AlCl_3} \xrightarrow{Zn-Hg/HCl}$$

(C)
$$CH_3$$
- CH_2 - CI CI_2 / $AICI_3$

(D)
$$Cl_2/Fe \rightarrow CH_3-CH_2-Cl$$
AlCl₃

The product C is:

The compound 'P' is:

(A)
$$CH_2$$
- CH_2 - $CH_$

10. Which of the following bromides is the major product of the reaction shown below, assuming that there are no carbocation rearrangement?

$$(A) \xrightarrow{+ HBr} (1 \text{ equivalent}) \xrightarrow{- C_{13}H_{17}Br} (B)$$

$$(B) \xrightarrow{- Br} (C)$$

$$(C) \xrightarrow{- Br} (D)$$

11. Which of the following reactions results in the formation of a pair of diastereomers?

(A)
$$\xrightarrow{\text{H}_3\text{C}} \xrightarrow{\text{HBr}}$$
 (B) $\xrightarrow{\text{H}_3\text{C}} \xrightarrow{\text{H}} \xrightarrow{\text{HBr}}$ (C) $\xrightarrow{\text{H}_3\text{C}} \xrightarrow{\text{H}} \xrightarrow{\text{HBr}} \xrightarrow{\text{HBr}}$

12. Which of the following compound was the starting material for the oxidation shown below?

$$(A) \qquad (B) \qquad (C) \qquad (D)$$

13. In which of the following reactions 1,3-butadiene will be obtained as a major product?

(A) Br – CH₂ – CH₂ – CH₂ – CH₂ – Br
$$\xrightarrow{\text{(CH3)3COK(2mole)}}$$

(B)
$$HO - CH_2 - CH_2 - CH_2 - CH_2 - OH \xrightarrow{conc.H_2SO_4}$$

(C)
$$H_2C = CH - C \equiv CH \xrightarrow{N_{i,R}} N_{i,R}$$

- (D) All of these
- **14.** Which combination is best for preparation of the compound (A) shown below?

(A)
$$H_3C$$
 $CH_2CH_2CH_2Br$ NaC CH_3CH_2 (A)

(B)
$$\stackrel{\text{CH}_3}{\underset{\text{CH}_3\text{CH}_2}{\longleftarrow}}$$
 (A)

15. In which reaction last product is $Ph - C \equiv CH$?

(A)
$$C_6H_5 - C - CH_3 \xrightarrow{3NaNH_2 \\ Mineral oil, heat} \xrightarrow{NH_4Cl}$$

(B)
$$C_6H_5CH = CH_2 \xrightarrow{CCl_4} \xrightarrow{3NaNH_2} \xrightarrow{NH_4Cl} \xrightarrow{NH_4Cl}$$

(C)
$$C_6H_5 - C - CH_3 \xrightarrow{PCl_5} \frac{3NaNH_2}{Mineral oil, heat} \xrightarrow{NH_4Cl}$$

(D) All

16. Identify the end product in the given sequence of reaction

$$\begin{array}{c} O \\ \hline \\ H_3C \end{array} \longrightarrow A \xrightarrow{H^+} A \xrightarrow{Me_2SO_4} B \xrightarrow{Conc.HNO_3} C$$

$$CH_3 \qquad CO.H.$$

(A)
$$OCH_3$$
 CO_2H

(B)
$$OCH_3$$
 CH_3

MULTIPLE CHOICE QUESTIONS

17. Which of the following compounds will give same major product on acid catalysed hydration

(A)
$$Ph - CH_3$$

 CH_3
 CH_3

$$C = C CH_3$$

$$C = C CH_3$$

$$CH_3$$

(D)
$$CH_3$$
 CH $C = CH_2$

18. Which of the following reactions are less possible by Friedel-craft's acylation as per expectation?

(A)
$$CH(CH_3)_2 \xrightarrow{CH_3CH_2COCI}$$

(B)
$$N(CH_3)_2 \xrightarrow{CH_3CH_2COCI}$$

(C)
$$CH_3CH_2COCI$$
AlCl₃

(D)
$$O_2N$$
 CH(CH₃)₂ CH_3CH_2COCI
AlCl₃

Compersion 19 to 21

Observe the following reaction and answer.

19. The compound (U) is (the major isomer is):

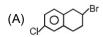
20. The compound (W) is

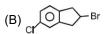






21. The compound (Z) is







22. Match the reaction of column (I) with the characteristics given in column (II)

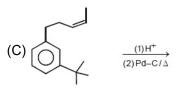
Column-I

(A) CH_3 $CONC.H_3PO_4$

- (p) Formation of carbocation intermediate
- (B) (1) AICI₃ (2) Pd-C/ Δ
- (q) Six membered ring closure in

Column-II

intramolecular reaction



(r) Aromatic Electrophilic substitution

(D) HCI (dry)

- (s) Dehydrogenation reaction
- (t) 5-membered ring closure in intramolecular reaction
- (A) (a) -p, q, r; (b) -p, q, r, s; (c) -p, q, r, s; (d) -p, r, t
- (B) (a) -p, q, r, s; (b) -p, q, r; (c) -p, q, s; (d) -p, r, s
- (C) (a) -p, q, s; (b) -p, q, r; (c) -p, q, r, t; (d) -p, q, t
- (D) (a) -p, q, t; (b) -p, q, r, t; (c) -p, q, r, s; (d) -p, s, t

SUBJECTIVE CHOICE QUESTIONS

23. Find out reagents involved in following conversions:

$$(a) \longrightarrow CH_3 - CH_2 - C - CH_3$$

$$(b) \longrightarrow CH_3 - CH_2 - CH_2 - C - H$$

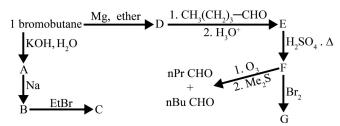
$$(c) \longrightarrow CH_3 - CH_2 - C - CH_3$$

$$(d) \longrightarrow CH_3 - CH_2 - C = C - CH_3$$

$$(e) \longrightarrow CH_3 - CH_2 - C = C - D$$

- **24.** C₅H₁₀ represents three isomeric alkenes A₁, A₂ and A₃. Each on hydrogenation gives 2-methylbutane. A₁ and A₂ on oxymercuration-demercuration give the same 3° alcohol. A₂ and A₃ on hydroboration oxidation give different 1° alcohol. Assign structures to A₁, A₂ and A₃ and explain the reactions.
- 25. One of the constituent of turpentine is α -pinene having molecular formula $C_{10}H_{16}$. The following scheme give reaction of α -pinene. Determine the structure of α -pinene & of the reaction products A through E.

26. Propose structures for intermediates & products A to G



27. What reagents would be required to carry out the following synthesis?

