

# **CHEMISTRY**

TARGET: JEE Advanced - 2023

# CAPS – 16 Alcohols, Phenols and Ethers

1. 
$$\xrightarrow{\text{CO}_2 / \text{NaOH}} \xrightarrow{\text{PhOH} / \text{H}_2 \text{SO}_4 / \Delta} \xrightarrow{\text{PhCOCI} / \text{Py}} (P)$$

Give the product (P) of the above reaction sequence :

$$(A) \overset{OH}{\smile} CHO \qquad (B) \overset{OH}{\smile} COOPh \qquad (C) \overset{OCOPh}{\smile} COOPh \qquad (D) \overset{OH}{\smile} COOPh$$

**2.** Identify 'Z' in the given sequence of reaction.

$$(A) \xrightarrow{\text{OH}} (B) \xrightarrow{\text{SOCI}_2} Y \xrightarrow{\text{AICI}_3} Z$$

3. Acetophenone  $\xrightarrow{HCO_3H}$  A  $\xrightarrow{H_3O^+}$  B + C  $\xrightarrow{Pthalic Anhydride}$  Indicator(D)

C & D are

(A) 
$$CH_3OH \& CH_3$$

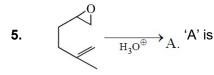
$$CH_3OH \& CH_3OH \& CH_3$$

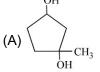
$$CH_3OH \& HO CH_3OH \& CH_3OH \& CH_3OH \& CH_3OH & CH_3OH$$

E is

4.

(A) 
$$CH_3$$
  $CH_2I$  (B)  $CH_2CH_3$  (C)  $CH_3$  (D)  $CH_3$ 





(D) 
$$CH_3$$
  $CH_3$ 

- **6.** Consider the reaction of HI with the following:
  - I O
- п

Which forms di-iodide on reaction with HI (excess)?

- (A) I and II both
- (B) II only
- (C) I only
- (D) none

7. 
$$\xrightarrow{\text{OH OH}} (A) \xrightarrow{\text{OH OH OH}} (B) \xrightarrow{\text{(1) MeMgBr}} (C) \xrightarrow{\text{NaBH}_4, \text{ EtOH}} (D)$$

Product (D) in above reaction is:

8. Select the correct synthesis products

(A) 
$$\xrightarrow{\text{BH}_3 \cdot \text{THF}} \xrightarrow{\text{H}_2\text{O}_2, \text{OH}^-} \xrightarrow{\text{NaOH}} \xrightarrow{\text{O}}$$

(B) 
$$\xrightarrow{\text{Hg(OAc)}_2} \xrightarrow{\text{NaBH}_4} \xrightarrow{\text{NaOH}} \xrightarrow{\text{O}}$$

$$(C) = \left( \begin{array}{c} \xrightarrow{\text{mCPBA}} \\ \xrightarrow{\text{CH}_2\text{Cl}_2} \end{array} \right)$$

(D) 
$$\left\langle \bigcirc \right\rangle$$
 -Br + (CH<sub>3</sub>)<sub>3</sub>CONa  $\xrightarrow{S_N 2}$  (CH<sub>3</sub>)<sub>3</sub>CO- $\left\langle \bigcirc \right\rangle$ 

- **9.** Which of the following are converted to aldehydes or ketones by MnO2?
  - (A)  $CH_2 = CH CHOH CH$

(B) 
$$\left\langle \begin{array}{c} \text{CH}_{3} \\ \text{OH} \end{array} \right\rangle$$

10. 
$$\begin{array}{c|c} Ph & CN & \\ H_{3}C & OH & \\ \hline & (A) & & \downarrow \\ & & Z & (A) \\ \end{array}$$

Correct statements are among the following

- (A) X gives yellow precipitate with I2 in presence of NaOH
- (B) Y is reduced by Fehling's solution
- (C) Z gives silver mirror with [Ag(NH<sub>3</sub>)<sub>2</sub>]<sup>+</sup>
- (D) X reduces Tollen's reagent.

- **11.** Which of the following is true for 3-methyl-2-butanone?
  - (A) It may be prepared by CrO<sub>3</sub> oxidation of 2-methyl-2-butanol.
  - (B) Its reaction with NaBH<sub>4</sub> gives a secondary alcohol.
  - (C) It may be prepared by acidic  $Hg^{2+}$  catalyzed hydration of 3-methyl-1-butyne
  - (D) This compound is an isomer of 4-penten-1-ol.

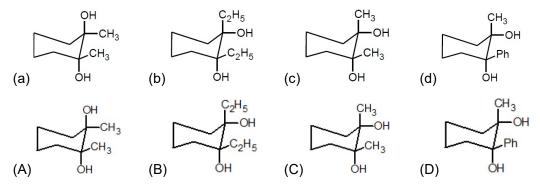
### Comprehension for 12 to 14

Pinacol is a 1,2, diol which on treating with acid produces pinacalone(ketone). It is an intramolecular rearrangement. The reaction starts with the protonation of hydroxyl group followed by elimination of water and formation of carbocation. The carbocation is then stabilized by With 1,2 shift.

12. 
$$\xrightarrow{OH} \xrightarrow{OH} \xrightarrow{OH} \xrightarrow{H^+} \text{Product (C) The product 'C' is}$$

$$(A) \begin{tabular}{c|c|c} \begin{tabular}{$$

**13.** Which of the following a compounds, on pinacol-pinacalone rearrangement produces a compound which gives a precipitate with KOI?



14. 
$$X \xrightarrow{Mg/THF} Y \xrightarrow{H^+} Ph - C - CH_3$$

$$CH_3O^+ \longrightarrow Ph - C - CH_3$$

Which of the following is not correct about the compound X and Y?

- (A) (X) is can be reduced to 1-phenylethane on treating with  $N_2H_4/C_2H_5O^-$
- (B) (X) on treating with H<sub>2</sub>N–OH / H<sup>+</sup> followed by treating with PCI<sub>5</sub> produces two amides.
- (C) (Y)- may be obtained by the crossed reductive coupling between acetone and benzophenone.
- (D) (X) gives yellow precipitate with NaOI.
- **15.** How many of the following methods would serve to prepare 1-phenyl-2-propanol?
  - (a) Addition of benzyl Grignard reagent to acetaldehyde (ethanal).
  - (b) Addition of phenyl lithium to propylene oxide (methyloxirane)
  - (c) Addition of phenyl Grignard reagent to acetone (2-propanone).
  - (d) Acid-catalyzed hydration (addition of water to) of 2—phenyl-1-propene.
  - (e) Addition of methyl Grignard reagent to acetophenone (methyl phenyl ketone).
  - (f) Addition of methyl Grignard reagent to phenylacetaldehyde.

16. Baeyer's reagent 
$$A \xrightarrow{HIO_4} B \xrightarrow{OH} A$$
 Products

The number of possible products is:

**17.** Match the column I with column II.

	Column-I (Reaction)		Column-II (Possible products)
(A)	$CH_3 \xrightarrow{\text{(i) Hg(OAc)}_2/\text{H}_2\text{O}}$	(p)	—CH <sub>2</sub> OH
(B)	$CH_2 \xrightarrow{\text{(i)}B_2H_0/\text{THF}}$	(q)	OH OH
(C)	$CH_3 \xrightarrow{H^9/H_2O}$	(r)	OH——CH <sub>s</sub>
(D)	$CH_3 \xrightarrow{\text{(i) Hg(OAc)}_2/\text{H}_2\text{O}}$	(s)	OH OH

(A) A 
$$\rightarrow$$
 q; B  $\rightarrow$  p; (C)  $\rightarrow$  r; (D)  $\rightarrow$  r

(B) A 
$$\rightarrow$$
 p; B  $\rightarrow$  q; (C)  $\rightarrow$  s; (D)  $\rightarrow$  r

(C) 
$$A \rightarrow q$$
;  $B \rightarrow p$ ; (C)  $\rightarrow r$ ; (D)  $\rightarrow s$ 

(D) 
$$A \rightarrow p$$
;  $B \rightarrow q$ ; (C)  $\rightarrow s$ ; (D)  $\rightarrow s$ 

### **18.** Match the Following:

Column I			Column II	
(A)	CH <sub>3</sub>	$\xrightarrow{1)H_2SO_4 \ conc.}$ $\xrightarrow{2)CH_3OH}$	(p)	CH <sub>3</sub>
(B)	CH <sub>3</sub>	1) <i>NaH</i> → 2) <i>CH</i> <sub>3</sub> <i>I</i> →	(p)	CH <sub>3</sub>
(C)	CH <sub>3</sub>	1) HBr 2)Mg 3)CH <sub>3</sub> I	(r)	CH <sub>3</sub>
(D)	OH OH	1) Na 2) CH <sub>3</sub> I	(s)	OCH <sub>3</sub>

(A) (A-s; B-r; C-q; D-p)

(B) (A-r; B-q; C-s; D-p)

(C) (A-p; B-r; C-q; D-q)

(D) (A-q; B-r; C-q; D-s)

## **Subjective**

- **19.** An organic compound 'A' having molecular formula C<sub>6</sub>H<sub>6</sub>O gives a characteristic colour with aqueous FeCl<sub>3</sub> solution. When 'A' is treated with CO<sub>2</sub> and NaOH at 400 K under pressure, 'B' is obtained. B on acidification gives C when C treated with CH<sub>3</sub>COCl gives a popular pain killer D. Deduce the structures of A, B, C and D.
- **20.** Complete the reaction equations :

- 21. Compound (A) C<sub>4</sub>H<sub>10</sub>O reacts rapidly with metallic sodium, but undergoes almost no reaction with Lucas reagent. When (A) is treated with hot concentrated sulphuric acid, a new compound (B) C<sub>4</sub>H<sub>8</sub> is formed. If C<sub>4</sub>H<sub>8</sub> is hydrated with sulphuric acid a new compound (C) C<sub>4</sub>H<sub>9</sub>OH is formed, which is almost inert to metallic sodium but reacts rapidly with Lucas reagent. What are (A), (B) and (C)?
- 22. In the following dehydration of diol with  $H_3PO_4$ , the following product is formed such that isotopic  $O^{18}$  goes with  $H_2O$ . Explain.

- 23. How can you convert PhCH = CH COCH $_3$  to
  - (i) PhCH =  $CHCO_2H$

(ii) PhCH = CHCH<sub>2</sub>CH<sub>3</sub>

(iii) PhCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>

(iv) PhCH = CHCH(OH)CH<sub>3</sub>

(v) PhCH<sub>2</sub>CH<sub>2</sub>COCH<sub>3</sub>